HETEROSUBSTITUTED ALLYLIC CARBANION BASED STEREOCONTROL, REGIO- AND STEREO-SELECTIVE REACTION OF 0 AND S SUBSTITUTED ALLYLIC CARBANIONS WITH ALDEHYDES

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Summary Reglo- and diastereoselective reaction of oxygen and sulfur substituted allyllc carbanlons with aldehydes IS described Either threo or erythro isomer can be obtained predominantly or exclusively by merely choosing an additive.

Diastereocontrol between two adjacent substituents in acyclic systems (<u>1</u>) continues to be of **a current importance in organic synthesis Such a stereochemical control between X = OH and Y = C IS realized through a number of methods 1 We now wish to report a convenient method for**

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\frac{1}{2} \quad x = 0H, Y = C, S_1, N, 0, S, Se,
$$

diastereocontrol between $X = OH$ and $Y = 0$ or between $X = OH$ and $Y = S^2$ Our method is based on regio- and stereoselective condensation between heterosubstituted allylic carbanions (2 or 3) **and aldehydes under the influence of an additive I'M"** erythro or threo diastereomer can be obtained by merely choosing "M" **sumnanzed in the Table. Our procedure IS very simple, either** The results are

Allyl Carbanion	Aldehyde	Additive "M"	Erythro, %	Threo, %
$\stackrel{2}{\equiv}$	PhCHO	Bu_3SnCl/BF_3	90	10
		Ph_3SnCl/BF_3	85	15
		Et ₂ A1CI	75	25
		Et ₃ A1	63	37
		Cp ₂ ZrCl ₂	$10\,$	90
	$(CH3)2$ CHCHO	Bu_3 SnC $1/BF_3$	-100	۰
$\overset{30}{=}$	PhCHO	Bu_3 SnCl/BF ₃	-100	
		-)-BC1	89	11
		EtAIC1 ₂	84	16
		Cp ₂ ZrCl ₂	27	73
$\underline{\underline{3b}}$	PhCHO	Et ₃ Al	92	$\bf 8$
		$Cp2$ Ti $C12$	$18\,$	82
	CH ₃ CH ₂ CHO	Et ₃ A1	-100	
	n-BuCHO	Et ₃ Al	-100	
	$CH_{3\text{N}}$ CH ₀ H	Et ₃ Al	-100	

Table Regio- and Stereoselective Coupling through Heterosubstituted Ally1 Carbanion ^a

^a All reactions were carried out on a l mmol scale as described in the text An appreciable amount of y-attacked product, YCH=CHCH₂CH(OH)R Y=S or 0, was not obtained, and total yields were in a range of 60 - 85 %. The ratio of erythro/threo was determined by GLPC and/or ¹H NMR **spectra** Erythro/threo nomenclature is based on Heathcock's convention

In the reaction of 2 , use of Bu₃SnC1/BF₃⁴ as an additive produces erythro isomer either predominantly or exclusively, while use of Cp₂ZrCl₂⁵ gives threo isomer 40 Other additives such as $\mathsf{Et}_3\mathsf{A1}^6$, $\mathsf{Et}_2\mathsf{A1}\mathsf{C1}$, and $\mathsf{R}_2\mathsf{B}\mathsf{C1}^7$ are not effective for stereocontrol, but interestingly they give erythro isomer preferentially This erythro preference is enhanced in the reaction of 3, boron and aluminum reagents produce erythro isomer with 84 - 100 % stereoselectivity Such a tendency is in marked contrast with the reaction of carbon \degree or silicon substituted allylic*'* **carbanlons, which afford threo isomer predominantly under the influence of boron and aluminum** reagents Presumably, "M" attacks the _Y-position of 2 or 3, producing an allylic organometallic compound like <u>4</u> Oxygen or sulfur atom strongly coordinates to M, so that the double bond is compound like 4

forced to take cis geometry IS **produced preferentially Such a coordination is** Consequently, erythro isomer Such a coordination is not **Y \ > & Y = 0 or S expected in Y = C or Si In the reaction of** $\frac{3b}{2}$ **, erythro isomer** is obtained with very high regio- and stereoselectivity even under the influence of Et₃Al which does not produce high stereoselectivity in other

reactions as shown in the Table

We next turned our attention to an application of the present method to the stereoselective synthesis of exo-brevlcomin. 9 Our route IS **summarized in the following scheme**

Treatment of MOM-allylic carbanion (3b) with Et₃Al at -78°C, followed by the addition of ketoaldehyde produced <u>5</u> (erythro/threo = 89/11)¹⁰ Acid treatment and subsequent reduction gave **exo-brevlcomlne (JJ -**

The general procedure is as follows To a solution of $\underline{\mathbf{2}}^{11}$ in dry ether was added an **additive** (1 **eq), and aldehyde** (1 **eq) was added after a few minutes Reaction mixture was** allowed to warm to room temperature, and then quenched with aqueous NH_4C1 solution at $0^{\circ}C$. Similar procedure was used for <u>3</u> Generation of <u>3g</u> is well known reactions, ¹² and 3 b was **generated by the reaction of ally1 methoxymethyl ether with n-BuLi (1 eq) in THF at -30°C**

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- 2) Diastereocontrol between $X = OH$ and $Y = 0$, see a) P G M. Wuts and S S Bigelow, J. Org **Chem , 47, 2498 (1982), b) R W. Hoffmann and B. Kemper, Tetrahedron Lett., 845 (1982), c) M. Yamaguchi and T. Mukalyama, Chem Lett** , **'237 (1982), d) T Mukaiyama and M Yamaguchi, lbld , 509 (1982), e) M. T. Reetz, R Steinbach, J Westermann, R Urz, B. Wendereth, and R Peter, Angew. Chem. Internat Ed** , Zl_, **135 (1982), f) C H Heathcock, J P Hagen, E. T. Jarvl, M C Plrrung, and S D. Young, J. Am Chem Sot , 103, 4972 (1981) Dlastereocontrol between X = OH and Y = S, see g) R W. Hoffmann and B Kemper, Tetrahedron Lett** , **4883 (1980). We also learned that Professor H Yamamoto succeeded stereocontrol between X = OH and Y = 0, or between X = OH and Y = S, as mentioned in the 9th round-table conference on organic reactions at Kyoto on August 2, 1982.**
- 3) Erytnro-, tnreo-isomer, and γ -adduct of 2 and $\overline{3}$ d were separated by GLPC (PEG 6000). The α adducts of 2 were converted to the corresponding sulfoxides through NaIO_A oxidation The structural determination is based on $\text{J}_{threo} \big\backslash \text{~J}_{ery\, throw}$ **assuming hydrogen bonding as shown in&and&** Stereoisomers from 3b were converted to vinyl epoxides, and then their structures were determined by ¹H NMR **pr!;&; cf&; H H** spectra, see J -C. Paladini and J. Chuche, Bull $\Phi = \sim 40^{\circ}$ $\Phi = \sim 160^{\circ}$ Soc Chim Fr., 187 (1974). Structure detern
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	- **7) These reagents produced threo Isomer either predominantly or exclusively in the reaction of trlmethylsllyl substituted allylic carbanlon**
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	- **9) When we were nearing completion of our synthesis, related synthesis via allyllc boronate derivative was reported (ref 2b).**
	- **10) The ratio was determined from the ratio of two MeO peaks in NMR spectra.** 6 **;** δ **(CCl₄) 1.35 (s,** 3H), 1 40-2 08 (m, 6H), 4 03 (bs, 1H), 4 24 (d, J=6, 1H), 4 96 (ddd, J=10.0, 2.0, and 1 0, 1H), 5.08 (ddd, J=16.5, 2 0, and 1 0, 1H), 5.69 (ddd, J=16 5, 10.0, and 6 0, 1H) \mathcal{I} ; δ (CC1_A) **0.87 (t, J=6 5, 3H),** 1 **28 (s, 3H), 1 00-2 00 (m, 8H), 3 76 (t, J=6 5, lH), 3 96 (bs,** 1H)
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	- **12) See ref 6 and references cited therein**

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